286. Asymmetric Induction in *Diels-Alder* Reactions to Acrylates Derived from Chiral sec-Alcohols

Preliminary Communication¹)

by Wolfgang Oppolzer, Mark Kurth, Daniel Reichlin, Christian Chapuis, Martin Mohnhaupt and Frank Moffatt

Département de Chimie Organique, Université de Genève, CH-1211 Genève 4

(11.XI.81)

Summary

Starting from the enantiomerically pure monoterpenes (+)-pulegone (3), (+)limonene (7), (-)- β -pinene (9), (+)- and (-)-camphor (13) or (+)-cholestenone (11) the chiral alcohols 4, 5, 6, 8, 10, 12, 14, 15, 16, 17 et 18 were prepared; their acrylates II underwent a TiCl₄-promoted *Diels-Alder* addition to cyclopentadiene (*Scheme 3*, *Table*) giving in a predictable manner either the (2R)- or the (2S)-adducts III with 63 to 88% asymmetric induction.

Further development of enantioselective C, C-bond-formation by *Diels-Alder* reactions remains a formidable challenge to the organic chemist despite recent progress in this field involving either chiral dienophiles [1] [2], dienes [3]²), or *Lewis*-acid catalysts [5]. Thus, the acrylate II of (-)-8-phenylmenthol (4), first reported by *Corey & Ensley* [2a], adds to cyclopentadiene in the presence of TiCl₄ to give the (2R)-norbornene III with ~90% asymmetric induction [6]. By analogy to the model proposed for intramolecular ene-type reactions of 8-phenylmenthyl enoates [7] this



¹) Presented by one of us (W.O.) at the 11th Northeast Regional Meeting of the American Chemical Society, Rochester, Oct. 19, 1981.

²) For an asymmetric intramolecular *Diels-Alder* reaction where the chirality-directing unit is attached to the chain which links the reaction partners see [4].

stereochemical control agrees with an acrylate conformation where the estercarbonyl group is antiplanar with the olefinic C, C-bond and synplanar with the alkoxy-C, H-bond (Scheme 1). Accordingly the phenyl ring of the ester unit shields the C(a)-re face by π , π -orbital overlap probably even more effectively in the presence of an appropriate Lewis acid³) thereby directing the diene addition to the dienophile-si-face. However, the need to purify the oily auxiliary alcohol (-)-4 by careful medium-pressure chromatography and the difficult access to the more interesting⁴) enantiomer (+)-4 seriously limits its applicability in organic synthesis. Nevertheless, the encouraging chiral induction observed with (-)-8-phenylmenthyl acrylate [2 a] [6], together with the above rationalisation served as a starting point for the design of more versatile and effective chiral auxiliary alcohols.

As an ultimate goal we aimed at the preparation and use of such alcohols which I) lead to quantitative asymmetric induction in addition reactions to their acrylates, 2) are easily accessible from inexpensive precursors in both antipodal forms or as an equivalent pair of *si*- and *re*-face directing isomers, 3) may be purified by crystallization and 4) are recovered easily after induction of chirality has been achieved. Apparent options include cyclohexanols which are locked in a rigid chair conformation with both equatorial hydroxyl group and an *a*-trans positioned arylsubstituted chain, as well as norbornanols carrying these crucial functionalities either in a *cis/endo* or *cis/exo* relationship. Accordingly, the *si*-face directing acrylates **A**, **B** and **C** and their *re*-face directing counterparts **D**, **E** and **F** may be schematically envisaged (Scheme 2). Our first steps towards this goal are sum-



marized in *Scheme 3* and in the *Table*. To prepare the cyclohexanols 4, 5, 6, 8, 10, 12 and the norbornanols 14, 15 and 18⁵) the steric position of the aryl-substituted chain was controlled by a base-induced equilibration of the corresponding cyclohexanones which then were reduced with sodium or lithium in 2-propanol [8].

³) In the ¹H-NMR. spectra the *trans*-H-C(β)-signal of 8-phenylmethyl acrylate is shifted upfield by 0.32 ppm in CDCl₃, and by 0.85 ppm in CDCl₃ containing 1 mol-equiv. of TiCl₄, compared to that of menthyl acrylate.

⁴) See for example the enantioselective synthesis of prostaglandins [2a].

⁵⁾ All new compounds were characterized by IR., ¹H-NMR. (360 MHz) and mass spectroscopy.

Starting material	Preparation a b,c	Auxiliary alcohol I R*OH, m.p. (°C)	Endo-adduct III Config. e.e.%	
Me 3 (+)-Pulegone		R ² R^{1} Ar $4 R^{1}=R^{2}=Me$, $Ar=Ph$ F $R^{1}=R^{2}=Me$, $Ar=pOMe-C_{6}H_{4}-$ $6 R^{1}=H$, $R^{2}=Ar=Ph$	(R) (R) (R)	89 87 88
(+)-L1monene	d	B Ph 90-91 8	(<i>R</i>)	63
9 (-)-8-Pinene	e,c	109.5	(<i>R</i>)	85
(+)- 4-Cholesten-3-one	f	HO Ph Ph Ph Ph 12	(S)	84
(+)-Camphor (+)- 13	i ع.h	14 R ¹ =H 47 Ph 15 R ¹ =Ph 173	(R) (R)	74 81.5
	١،٩	Holin oil	No Re	action
II	K	Ar OH oil 17	(5)	88
(-)-Camphor (-)- 13	ç, h	Ph H 173	(S)	82.7

Table. Preparation of the chiral alcohols I and asymmetric induction in cycloadditions of their acrylates $II \rightarrow III$

Preparation a. 1) (p-OMe-C₆H₄)₂CuLi (1 mol-equiv. CuBr · Me₂S, 2 mol-equiv. (p-OMe-C₆H₄)₂Li, prepared from p-OMe-C₆H₄Br and BuLi at -20°) ether, $-20^{\circ} \rightarrow$ RT., 48 h, 21%; 2) slow addition in 8 mol-equiv. 2-propanol to 6 mol-equiv. sodium metal in toluene at reflux, 1 h reflux, 48%.

Preparation b. 1) 8% H₂SO₄ in EtOH/H₂O 2.5:1, reflux 15 h, 65% [9]; 2) 1.05 mol-equiv. PhCHO, 4% aq. KOH-solution, 3 h, 86% [10]; 3) 3 mol-equiv. PhLi, 2 mol-equiv. CuBr · Me₂S, Me₂S/ether 1:2, 1.5 h at -20° , 1 h at 0° , 46%.

Preparation c. 5 Mol-equiv. sodium metal, 10 mol-equiv. 2-propanol, toluene, 0.5 h at RT., 46%.

Preparation d. 1) 1 Mol-equiv. MCPBA, CH_2Cl_2 , 1.5 h at 5°, 90%; 2) Pt, I atm. H₂, 72% [11]; 3) 1.4 mol-equiv. diethylaluminium 2,2,6,6-tetramethylpiperidide [12], benzene, 80 min at 0°, 88%; 4) 1.1 mol-equiv. DMSO, 1.1 mol-equiv. (COCl)₂, excess NEt₃, CH_2Cl_2 , $-78^\circ \rightarrow RT.$, 93% [13]; 5) Ph₂CuLi (2.85 mol-equiv. PhLi, 2 mol-equiv. CuBr, THF, 1 h at -20° , 45 min at -78° , 45 min at -20° , 90 min at RT., 30%; 6) 0.25N NaOH, EtOH, 70°, 2 h, 93%; 7) 5 mol-equiv. sodium metal, 1.2 mol-equiv. 2-propanol/toluene, 2 h at reflux, 38%, m.p. 90–91° (pentane).

Preparation e. 1) O₃, -70° , MeOH, Me₂S, 81% [14]; 2) PhCHO, NaOMe, MeOH, 78% [15]; 3) Ph₂CuLi (3.7 mol-equiv. Cul, 5.8 mol-equiv. PhLi, THF, 0°, 20 min) $-20^{\circ} \rightarrow RT.$, 87%, m.p. 108–109° (hexane/ether); 73%.

Preparation f. 1) 2 Mol-equiv. t-BuOK, 1.37 mol-equiv. Ph₂CHCl, t-BuOH, reflux, 1 h; 2) Excess sodium metal, 30 mol-equiv. 2-propanol/toluene, RT., 28% overall.

Preparation g. 1.1 Mol-equiv. sodium metal, benzene, reflux, 18 h, 1.5 mol-equiv. PhCHO, 4 h at RT., 55% [16]; 2) 1 Atm. H₂, Pd/C 5%, MeOH, 4 h at RT., 80%; 3) 1.2N NaOMe, MeOH, 24 h at 80°, 85%.

Preparation h. 2.2 Mol-equiv. lithium metal, NH₃/EtOH/Et₂O 6.8:1.8:1, 2 h, 61% [17], m.p. 46-47° (pentane).

Preparation i. 1) 2 Mol-equiv. $CF_3SO_3SiMe_3$, 2.5 mol-equiv. NEt_3 , $0^\circ \rightarrow RT$., 2 h, 94% [18]; 2) 0.9 mol-equiv. Ph₂CHCl, 1 mol-equiv. TiCl₄ - 78° $\rightarrow RT$. [19], 48%, m.p. 132°; 3) 1 mol-equiv. LiAlH₄, ether, 1 h at RT., chromatography (SiO₂), 20%.

Preparation j. 4 Mol-equiv. LiAlH₄, ether, reflux, 4 h, 55% overall (g+j).

Preparation k. 1) 1.73 Mol-equiv. SeO₂, Ac₂O, reflux, 6 h, 94% [20]; 2) H₂, Ra-Ni, 1 atm. H₂, EtOH, 1 h, 86% [21]; 3) 1 mol-equiv. NaH, 1.3 mol-equiv. PhCH₂Cl, DMF, 5 h at -40° , 6 h at RT., 56%; 4) 1.16 mol-equiv. L-selectride [22], THF, 2 h at -78° , 1 h for \rightarrow RT, 88%.



After esterification⁶) the acrylates II (0.925 mmol) were treated with TiCl₄ (1.4 mmol) in dry CH₂Cl₂ (10 ml) at 0° for 45 min. After slow addition of 1.21 m freshly distilled cyclopentadiene in CH₂Cl₂ (2.78 mmol) the mixture was stirred at 0° for 4 h. Work-up gave the cycloadducts III which were isolated, purified and analyzed as described previously [6] by conversion to 1 and 2 with recovery of the auxiliary alcohol I.

First, the (-)-8-*p*-methoxyphenylmenthol $(5)^5$) was prepared in close analogy to (-)-4 from (+)-pulegone (3).

We hoped that an increased π -orbital interaction between the *p*-methoxyphenyl and the C(*a*), C(β)-bond of the enoate moiety in the acrylate of 5 would lead to a

⁶) The esters II⁵) were obtained by treatment of the alcohols I with acryloyl chloride (2 mol-equiv.), NEt₃ (2 mol-equiv.), *p-N*, *N*-dimethylaminopyridine (0.14 mol-equiv.) in CH₂Cl₂ at 0° for 1 h in 52 to 93% yield.

higher optical yield; this did not materialize and the adduct III was obtained in 87% e.e.⁷).

More encouraging was the induction observed on *Diels-Alder* addition to the acrylate of 6 (88% e.e.) which indicated an equally powerful enoate shielding of the dimethylphenylmethane and the diphenylmethane units. The latter moiety is readily introduced either by phenylcuprate addition to *a*-benzylidenecyclohexanones or by alkylation of dienolates and of silyl vinyl ethers (in the presence of TiCl₄ [19]) with diphenylchloromethane as illustrated by the preparation of the alcohols 6, 10, 12, 15 and 18. In comparison, a simple benzyl group shields less efficiently, leading to 63 and 74% chiral induction in the addition of cyclopentadiene to the acrylates of 8 and 14, respectively. However, the alcohols 4, 5 and 6 display two disadvantages: lack of crystallinity and the commercial availability only of the (+)-enantiomer of the starting pulegone (3). By contrast, both antipodes of limonene (7) are inexpensive as well as $(-)-\beta$ -pinene (9)⁸).

Thus, the monoterpenes 7 and 9 served as convenient precursors to the crystalline alcohols 8 and 10; the latter shows a chiral auxiliary capacity comparable to the reference alcohol 4. Recent access to steroids⁹) renders this rigid skeleton an attractive starting material. For example, cholestenone (11) was transformed by two steps into the alcohol 12; its acrylate gave the (2S)-cyclopentadieneadduct III with 84% e.e. and thus constitutes the first, readily available 're-face directing' acrylate.

For the preparation of the envisaged chiral norbornanols camphor (13) is a particularly suitable chiral source for the following reasons: 1) both pure enantiomers are easily available: (+)-13 commercially and (-)-13 by Jones' oxidation [26] of (-)-borneol (Aldrich, EGA); 2) substituents may be directed into the more stable exo- or into the endo-position by means of thermodynamically vs. kinetically controlled processes; 3) derivatives of 13 are frequently crystalline. Thus, the crystalline endo/cis-benzyl- and diphenylmethyl-isoborneols 14 and 15⁵) were obtained from (+)-camphor to afford after esterification⁶) and subsequent Diels-Alder reaction the (2R)-adducts III with 74 and 81.5% asymmetric induction, respectively. The acrylate of the exo/cis-benzylborneol afforded no Diels-Alder products but led to polymerization under more stringent reaction conditions. On the other hand, the regio- and stereo-selectivity prepared exo/cis-benzyloxyborneol 17¹⁰) gave an acrylate which yielded smoothly the (2S)-adduct III in 88% e.e. Accordingly the alcohol 17 constitutes a further efficient 're-face directing' auxiliary as well as the norbornanol 18 which was prepared from (-)-camphor.

Presently we are pursuing these leads further to arrive at new chiral acrylates which meet all the requirements for their generally practical use in enantioselective *Diels-Alder*, ene- and 1,4-addition reactions¹¹).

⁷⁾ The electron-donating effect of the p-OMe group may be compensated by its association with TiCl₄.

⁸⁾ Commercially available (+)-a-pinene may be converted to (+)-β-pinene following established procedures [23]. For stereoselective functionalisations of pinenes see a review [24].

⁹) See for example the fermentation of β -sitosterol [25].

¹⁰) For a non-selective preparation of **17**, requiring its isolation by HPLC. and for the asymmetric *Diels-Alder* addition of its fumarate to anthracene see [2c].

¹¹) For highly enantioselective BF₃-mediated organocuprate additions to enoates derived from (-)-8-phenylmenthol see the subsequent communication [27].

Financial support of this work by the Schweizerischer Nationalfonds zur Förderung der wissenschaftlichen Forschung, Sandoz AG, Basel, and Givaudan SA, Vernier, is gratefully acknowledged. We also thank Mr. J. P. Saulnier, Mr. A. Pinto and Mrs. D. Clément for NMR. and MS. measurements.

REFERENCES

- [1] a) H. M. Walborsky, L. Barash & T. C. Davis, Tetrahedron 19, 2333 (1963); b) J. Sauer & J. Kredel, Tetrahedron Lett. 1966, 6359; c) R. F. Farmer & J. Hamer, J. Org. Chem. 31, 2418 (1966).
- [2] a) E.J. Corey & H.E. Ensley, J. Am. Chem. Soc. 97, 6908 (1975); b) R.K. Boeckman, jr., P.C. Naegely & S.D. Arthur, J. Org. Chem. 45, 752 (1980); c) G. Helmchen & R. Schmierer, Angew. Chem. Int. Ed. 20, 205 (1981); Angew. Chem. 93, 208 (1981).
- [3] a) B.M. Trost, S.A. Godleski & J.P. Genêt, J. Am. Chem. Soc. 100, 3930 (1978); b) B.M. Trost,
 D. O'Krongly & J.L. Belletire, J. Am. Chem. Soc. 102, 7595 (1980); c) S. David, J. Eustache & A. Lubineau, J. Chem. Soc., Perkin I, 1979, 1795.
- [4] T. Mukaiyama & N. Iwasawa, Chem. Lett. 1981, 29.
- [5] S. Hashimoto, N. Komeshima & K. Koga, J. Chem. Soc. Chem. Commun. 1979, 437.
- [6] W. Oppolzer, M. Kurth, D. Reichlin & F. Moffatt, Tetrahedron Lett. 1981, 2545.
- [7] W. Oppolzer, C. Robbiani & K. Bättig, Helv. Chim. Acta 63, 2015 (1980).
- [8] H.O. House, 'Modern Synthetic Reactions', 2nd Ed., W.A. Benjamin, 1972, p. 150.
- [9] H. Rupe, Ann. 459, 195 (1927), E.J. Eisenbraun & S.M. McElvain, J. Am. Chem. Soc. 77, 3383 (1955).
- [10] W. C. M. C. Kokke & F.A. Varkevisser, J. Org. Chem. 39, 1535 (1974).
- [11] S. G. Traynor, B. J. Kane, M. F. Betkouski & L. M. Hirschy, J. Org. Chem. 44, 1557 (1979).
- [12] A. Yasuda, S. Tanaka, K. Oshima, H. Yamamoto & H. Nozaki, J. Am. Chem. Soc. 96, 6513 (1974).
- [13] A.J. Mancuso, S.-L. Huang & D. Swern, J. Org. Chem. 43, 2480 (1978).
- [14] J. Grimshaw, J. T. Grimshaw & H. R. Juneja, J. Chem. Soc., Perkin 1, 1972, 50.
- [15] O. Wallach, Justus Liebigs Ann. Chem. 313, 363 (1900).
- [16] J.C. Richer & A. Rossi, Can. J. Chem. 50, 1376 (1972).
- [17] J. W. Huffmann & W. W. Mc Whorter, J. Org. Chem. 44, 594 (1979).
- [18] G. Simchen & W. Kober, Synthesis 1976, 259.
- [19] M.T. Reetz & W.F. Maier, Angew. Chem. 90, 50 (1978); Angew. Chem. Int. Ed. Engl. 17, 48 (1978).
- [20] H. Rupe & A.T. di Vignano, Helv. Chim. Acta 20, 1078 (1937).
- [21] B. Pfrunder & Ch. Tamm, Helv. Chim. Acta 52, 1630 (1969).
- [22] H.C. Brown & S. Krishnamurthy, J. Am. Chem. Soc. 94, 7159 (1972).
- [23] L. M. Harwood, M. Julia, Synthesis 1980, 456; G. L. Kaiser (SCM Corp.) U.S.
- [24] D. V. Banthorpe and D. Whittaker, Chem. Rev. 66, 643 (1966).
- [25] a) C.K.A. Martin, Adv. Appl Microbiol. 22, 29 (1977); b) K. Kieslich, Ann. Rep. Fermentation Procedures, Academic Press 1978, 369; c) M.G. Wovcha, F.J. Antosz, J.C. Knight, L.A. Kominek, T.R. Pyke, Biochim. Biophys. Acta 531, 308 (1978).
- [26] Fieser & Fieser, 'Reagents for organic synthesis', Wiley Interscience Vol. I, 1967, 142.